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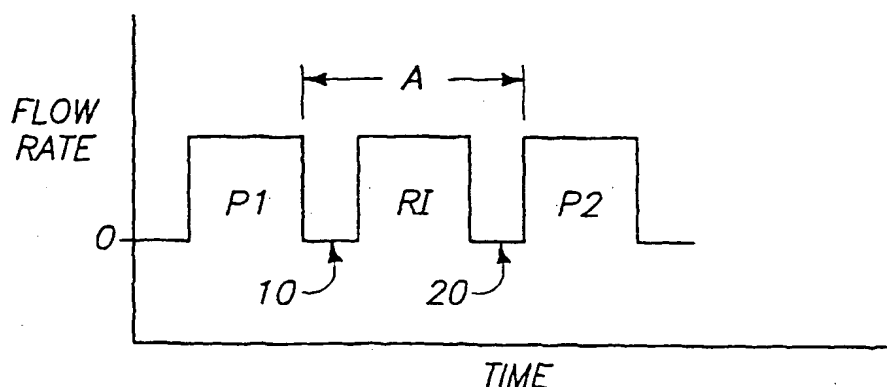
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(54) Title: ATOMIC DEPOSITION LAYER METHODS



(57) Abstract: An atomic layer deposition method includes positioning a semiconductor substrate within an atomic layer deposition chamber. A first precursor gas is flowed to the substrate within the atomic layer deposition chamber effective to form a first monolayer on the substrate. After forming the first monolayer, a reactive intermediate gas is flowed to the substrate within the deposition chamber. The reactive intermediate gas is capable of reaction with an intermediate reaction by-product from the first precursor flowing under conditions of the reactive intermediate gas flowing. After flowing the reactive intermediate gas, a second precursor gas is flowed to the substrate within the deposition chamber effective to form a second monolayer on the first monolayer. Other aspects and implementations are contemplated.

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**DESCRIPTION**  
**ATOMIC DEPOSITION LAYER METHODS**

**Technical Field**

5           This invention relates to atomic layer deposition methods.

**Background Art**

Atomic layer deposition (ALD) involves the deposition of successive monolayers over a substrate within a deposition chamber typically maintained at subatmospheric pressure. An exemplary such method includes feeding a single  
10 vaporized precursor to a deposition chamber effective to form a first monolayer over a substrate received therein. Thereafter, the flow of the first deposition precursor is ceased and an inert purge gas is flowed through the chamber effective to remove any remaining first precursor which is not adhering to the substrate from the chamber. Subsequently, a second vapor precursor different from the first is flowed  
15 to the chamber effective to form a second monolayer on/with the first monolayer. The second monolayer might react with the first monolayer. Additional precursors can form successive monolayers, or the above process can be repeated until a desired thickness and composition layer has been formed over the substrate.

It is a desired intent or effect of the purging to remove gas molecules that  
20 have not adsorbed to the substrate or unreacted gas or reaction by-products from the chamber to provide a clean reactive surface on the substrate for the subsequent precursor. In the context of this document, a reaction by-product is any substance (whether gas, liquid, solid or mixture thereof) which results from reaction of any deposition precursor flowing to the chamber and that is not desired to be deposited  
25 on the substrate. Further in the context of this document, an intermediate reaction by-product or reaction intermediate by-product is a reaction by-product formed by less than complete reaction of a precursor to form a desired monolayer on the substrate. Where there is a great degree of varying topography and/or there are high aspect ratio features on the substrate, it can be difficult to move the unreacted  
30 gases or reaction by-products from deep within openings for ultimate removal from the chamber. Further, certain reaction by-products, particularly intermediate reaction by-products, may not be gaseous and may not completely react to form gaseous reaction by-products in the typical short precursor pulse times. Accordingly, the purge gas pulse may not be effective or sufficient in removing such intermediate  
35 reaction by-products from the substrate and chamber.

For example, consider that in an atomic layer deposition of titanium nitride using  $\text{TiCl}_4$  and  $\text{NH}_3$ , the desired deposition product is  $\text{TiN}$  with  $\text{HCl}$  gas being the desired principle gaseous by-product. Consider also that there might be reaction intermediate by-products which might, even if gaseous, be difficult to remove from substrate openings. Further, if certain reaction intermediate by-products are solid and/or liquid phase prior to  $\text{HCl}$  formation, complete removal can be even more problematic where less than complete reaction to  $\text{TiN}$  and  $\text{HCl}$  occurs.

Consider also the atomic layer deposition of  $\text{Al}_2\text{O}_3$  using trimethylaluminum (TMA) and ozone as alternating deposition precursors. Apparently in such deposition, achieving an effective ozone precursor feed can be somewhat of a challenge due to the limited lifetime of ozone within the chamber. Specifically, an ozone molecule is in an inherently unstable, reactive form of oxygen which can rapidly dissociate and/or combine with another ozone molecule to form three  $\text{O}_2$  molecules. Regardless, a desired goal in the ozone feed is adsorption of oxygen atoms from the  $\text{O}_3$  to the surface of the substrate with  $\text{O}_2$  as the reaction by-product which is driven off. Of course, the  $\text{O}_2$  which forms deep within openings on the substrate has to be removed therefrom while more  $\text{O}_3$  needs to get into the openings to form a complete monolayer of oxygen atoms adhering to the substrate. In other words, the  $\text{O}_2$  which forms is trying to get out while more  $\text{O}_3$  is trying to get in.

While the invention was motivated in addressing the above issues and improving upon the above-described drawbacks, it is in no way so limited. The invention is only limited by the accompanying claims as literally worded (without interpretative or other limiting reference to the above background art description, remaining portions of the specification or the drawings) and in accordance with the doctrine of equivalents.

### Summary

The invention includes atomic layer deposition methods. In one implementation, an atomic layer deposition method includes positioning a semiconductor substrate within an atomic layer deposition chamber. A first precursor gas is flowed to the substrate within the atomic layer deposition chamber effective to form a first monolayer on the substrate. After forming the first monolayer, a reactive intermediate gas is flowed to the substrate within the deposition chamber. The reactive intermediate gas is capable of reaction with an intermediate reaction by-product from the first precursor flowing under conditions of the reactive intermediate gas flowing. After flowing the reactive intermediate gas, a second

precursor gas is flowed to the substrate within the deposition chamber effective to form a second monolayer on the first monolayer.

In one implementation, an atomic layer deposition method includes positioning a semiconductor substrate within an atomic layer deposition chamber. A first precursor gas is flowed to the substrate within the atomic layer deposition chamber effective to form a first monolayer on the substrate. After forming the first monolayer, a plurality of inert purge gas pulses are flowed to the substrate within the deposition chamber, with the plurality of inert purge gas pulses including at least some period of time between at least two adjacent inert purge gas pulses when no gas is fed to the chamber. After the plurality of inert purge gas pulses, a second precursor gas is flowed to the substrate within the deposition chamber effective to form a second monolayer on the first monolayer.

Other aspects and implementations are contemplated.

#### **Brief Description of the Drawings**

Preferred embodiments of the invention are described below with reference to the following accompanying drawings.

Fig. 1 is a diagrammatic depiction of flow versus time of one atomic layer deposition process in accordance with an aspect of the invention.

Fig. 2 is a diagrammatic depiction of flow versus time of one atomic layer deposition process in accordance with an aspect of the invention.

Fig. 3 is a diagrammatic depiction of flow versus time of one atomic layer deposition process in accordance with an aspect of the invention.

Fig. 4 is a diagrammatic depiction of flow versus time of one atomic layer deposition process in accordance with an aspect of the invention.

Fig. 5 is a diagrammatic depiction of flow versus time of one atomic layer deposition process in accordance with an aspect of the invention.

#### **Best Modes for Carrying Out the Invention and Disclosure of Invention**

An atomic layer deposition method in accordance with an aspect of the invention includes positioning a semiconductor substrate within an atomic layer deposition chamber. In the context of this document, the term "semiconductor substrate" or "semiconductive substrate" is defined to mean any construction comprising semiconductive material, including, but not limited to, bulk semiconductive materials such as a semiconductive wafer (either alone or in assemblies comprising other materials thereon), and semiconductive material layers (either alone or in assemblies comprising other materials). The term "substrate" refers to any supporting structure, including, but not limited to, the semiconductive substrates described above.

A first precursor gas is flowed to the substrate within the atomic layer deposition chamber effective to form a first monolayer on the substrate. Any first precursor gas is contemplated, whether existing or yet-to-be developed. By way of example only, where a desired ultimate deposition product or layer is TiN, an example first precursor could be either  $\text{TiCl}_4$  or  $\text{NH}_3$ . Further by way of example only, where the ultimate layer or product being formed is to be  $\text{Al}_2\text{O}_3$ , exemplary first precursor gases include trimethylaluminum and ozone. By way of example only, an exemplary first monolayer utilizing  $\text{TiCl}_4$  would include titanium or a titanium complex, whereas with respect to  $\text{NH}_3$  such would at least include nitrogen. With respect to trimethylaluminum, the first monolayer would include an aluminum complex, and with ozone typically adhered oxygen atoms. Any suitable temperature, pressure, flow rate or other operating parameters, with or without plasma, can of course be selected and optimized by the artisan with no particular set of the same being preferred or constituting a part of the invention.

After forming the first monolayer, a reactive intermediate gas is flowed to the substrate within the deposition chamber. In the context of this document, a "reactive intermediate gas" is one which is capable of reaction with an intermediate reaction by-product from the first precursor flowing under conditions of the reactive intermediate gas flowing. Further and preferably, the reactive intermediate gas is not capable of reaction with either the first monolayer under the reactive intermediate gas flowing conditions, nor capable of reaction with any of the first precursor which might be remaining in the reactor under the reactive intermediate gas flowing conditions. Further in one preferred embodiment, the reactive intermediate gas is not capable of reaction with, and does not react with, any exposed portion of the substrate under the reactive intermediate gas flowing conditions during such flowing of the reactive intermediate gas.

In one aspect, the invention contemplates reactive intermediate gas flowing regardless of whether any intermediate reaction by-product actually forms during the first precursor gas flowing. Further in another preferred aspect, the invention also contemplates forming some intermediate reaction by-product during the first precursor gas flowing, and then reacting the reactive intermediate gas with the intermediate reaction by-product during the reactive intermediate gas flowing.

The conditions (i.e., temperature, pressure, flow rate, etc.) of the reactive intermediate gas flowing to the substrate within the deposition chamber can be optimized by the artisan and are not otherwise particularly germane or preferred to any aspect of the invention. By way of example only, such conditions might be

the same as or different from any of a first precursor gas flow, a second precursor gas flow and/or inert purge gas flow. In one preferred aspect, the reactive gas flowing is plasma enhanced for example by one or both of plasma generation within the chamber and plasma generation remote of the chamber.

5       The particular reactive intermediate gas selected, whether a single constituent or a mixture of constituents, will depend as a minimum upon at least one intermediate reaction by-product from the first precursor flowing with which the reactive intermediate gas would be capable of reaction under conditions of the reactive intermediate gas flowing. By way of example only, two possible  
10       intermediate reactive gas components include  $\text{Cl}_2$  and  $\text{H}_2$ , and which would be expected to be particularly useful with one or both of  $\text{TiCl}_4$  and  $\text{NH}_3$  precursors. For example, with respect to  $\text{TiCl}_4$  and  $\text{NH}_3$  reaction by-products, such might include  $\text{TiCl}_2$ ,  $\text{TiCl}_3$  and  $\text{NH}_3$  complexes.  $\text{Cl}_2$  and/or  $\text{H}_2$  could be provided to the reactor for availability to react with these reaction by-products to facilitate their  
15       conversion to the ultimate stoichiometric equation reaction by-product (i.e., gaseous  $\text{HCl}$ ), or react to form some other product which might be more easily swept from the reactor either by the flowing reactive intermediate gas, by some other gas or by no other gas flowing.

After flowing the reactive intermediate gas, a second precursor gas is flowed  
20       to the substrate within the deposition chamber effective to form a second monolayer on the first monolayer, with "on" of course meaning in at least partial direct physical contact with the first monolayer. The second precursor is typically and preferably different in composition from the first precursor. Further preferably, the reactive intermediate gas is chosen to not be capable of reaction with the second  
25       precursor under the reactive intermediate gas flowing conditions. By way of example only, exemplary second precursors include any of the above  $\text{TiCl}_4$ ,  $\text{NH}_3$ , TMA and ozone in the respective exemplary deposition reactions referred to above. In accordance with any existing or yet-to-be developed atomic layer deposition method, the formed second monolayer might combine with the first monolayer to  
30       thereby inherently form a desired deposition product (i.e.,  $\text{TiN}$  or  $\text{Al}_2\text{O}_3$ , with the above referred to respective precursors), or to form a second monolayer that does not necessarily inherently react with the underlying first monolayer. The particular conditions of the flowing of the second precursor can be optimized by the artisan, of course, and do not constitute a material or preferred aspect of the inventions  
35       disclosed herein.

By way of example only, Fig. 1 depicts an exemplary plot of flow rate versus time of but one process in accordance with one aspect of the invention. In Fig. 1, P1 refers to a time period during which a first precursor flows to the chamber; RI refers to a time period during which a reactive intermediate gas flows to the substrate; and P2 refers to a time period during which a second precursor flows to the chamber. Accordingly, Fig. 1 depicts a preferred embodiment wherein the first precursor gas flowing, the reactive intermediate gas flowing and the second precursor gas flowing collectively comprise multiple time-spaced gas pulses to the substrate within the deposition chamber. In one preferred embodiment, there is some period of time between the first precursor gas flowing and the reactive intermediate gas flowing during which no gas is fed to the chamber, and as illustrated by example only with respect to location 10. Further in one preferred embodiment, there is a period of time between the reactive intermediate gas flowing and the second precursor gas flowing during which no gas is fed to the chamber and as, by way of example only, is indicated with the numeral 20. The particular length and rates of the respective flowings, and times therebetween, can be optimized by the artisan, of course, and do not constitute a material or preferred aspect of the inventions disclosed herein. Further, a first precursor gas pulse and a second precursor gas pulse are only stated or related temporally, of course, and not necessarily that the first stated precursor gas flowing or pulsing is a first ever precursor flowing to the chamber. Likewise and accordingly depending upon which portion of a time line much like the figures to which one is looking, second precursor gas pulses can be considered as first precursor gas pulses, and vice versa.

Fig. 1 also depicts one embodiment whereby there exists a total period of time "A" between first precursor feeding P1 and second precursor feeding P2. Total period of time A is characterized by no gas being fed to the chamber during said total period but for said reactive intermediate gas flowing RI. However, the invention also contemplates other gas flowings intermediate a first precursor feeding and a second precursor feeding. By way of example only, such are depicted in Figs. 2-5 in certain aspects. Fig. 2 depicts the flowing of an inert purge gas in the form of an inert gas pulse (IN) which is time-spaced from all other gas flowings to the substrate within the deposition chamber after flowing reactive intermediate gas pulse RI. Accordingly by way of example only, the exemplary diagrammatic Fig. 2 processing depicts a total period of time "B" between first precursor feeding P1 and second precursor feeding P2 having more than just the reactive intermediate gas flow RI.

Fig. 3 depicts another embodiment similar to Fig. 2, but for the reversal of the sequence of the inert gas pulsing and reactive intermediate gas pulsing between the first and second precursor flows.

Fig. 4 depicts another exemplary atomic layer deposition method which, in certain aspects, is not dependent upon any reactive intermediate gas flowing as described above. Specifically and by way of example only, Fig. 4 depicts, after the forming of a first monolayer from a precursor P1 flowing, a plurality of inert purge gas pulses IN flowing to the substrate within the deposition chamber, with such inert purge gas pulses being time-spaced such that there is at least some period of time (i.e., a period 50) between at least two adjacent inert purge gas pulses IN when no gas is fed to the chamber. By way of example only and in no way of limitation, such multiple, time-spaced inert purge gas pulses might facilitate the removal of reaction intermediate by-products, particularly from within high aspect ratio openings on the substrate. In the context of this document, "two adjacent inert purge gas pulses" refers to two inert purge gas pulses that have no other purge gas pulse therebetween. The phrase "two immediately adjacent inert purge gas pulses" refers to two inert purge gas pulses that have no gas pulse therebetween.

Fig. 4 depicts an exemplary embodiment wherein the plurality of inert purge gas pulses intermediate the first and second precursor flowings is two in number. Of course, the number of inert purge gas pulses to the chamber might be more than two in number. Further, the inert purge gas pulses might be of the same or common inert purge gas composition within the illustrated or all of the plurality of purge gas pulses, or be of some degree of varying composition. Fig. 4 also depicts but one preferred embodiment wherein there is a total period of time "C" between the first precursor feeding and the second precursor feeding. Total period of time C is characterized by no gas being fed to the chamber during such total period but for the plurality of inert purge gas pulses IN. Further considered and by way of example only, the depicted exemplary Fig. 4 processing illustrates a total period of time "D" between the illustrated two immediately adjacent inert purge gas pulses IN. Such total period of time D is characterized by no gas being fed to the chamber during such total period D.

By way of example only, Fig. 5 illustrates an exemplary alternate process having a total period of time D' between the two adjacent inert purge gas pulses IN which is characterized by or includes a reactive intermediate gas flowing RI to the substrate within the deposition chamber, for example in accordance with the



first-described embodiments. Accordingly and by way of example only, Fig. 5 also depicts in accordance with the first-described embodiments, the flowing of multiple time-spaced inert purge gas pulses IN to the substrate within the deposition chamber intermediate the first precursor flowing and the second precursor flowing. Of course, the IN and RI pulses could be ordered in any conceivable manner between the first and second precursor flowings, as well as other inert gas pulses, reactive intermediate gas pulses or other gas pulses added between the first and second precursor flows P1 and P2, respectively. Further, any processing of the substrate could occur, either before or after the Figs. 1-5 depicted processings, and the depicted processings would typically and preferably be repeated as selected by the artisan to achieve a desired thickness deposition by atomic layer deposition on the substrate.

The invention might have particular applicability to formation of hemispherical grain polysilicon container capacitors formed within high aspect ratio openings, but is in no way so limited.

In compliance with the statute, the invention has been described in language more or less specific as to structural and methodical features. It is to be understood, however, that the invention is not limited to the specific features shown and described, since the means herein disclosed comprise preferred forms of putting the invention into effect. The invention is, therefore, claimed in any of its forms or modifications within the proper scope of the appended claims appropriately interpreted in accordance with the doctrine of equivalents.

CLAIMS

1. An atomic layer deposition method, comprising:  
positioning a semiconductor substrate within an atomic layer deposition  
5 chamber;  
flowing a first precursor gas to the substrate within the atomic layer  
deposition chamber effective to form a first monolayer on the substrate;  
after forming the first monolayer, flowing a reactive intermediate gas to the  
substrate within the deposition chamber, the reactive intermediate gas being capable  
10 of reaction with an intermediate reaction by-product from the first precursor flowing  
under conditions of the reactive intermediate gas flowing; and  
after flowing the reactive intermediate gas, flowing a second precursor gas  
to the substrate within the deposition chamber effective to form a second monolayer  
on the first monolayer.  
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2. The method of claim 1 comprising forming the intermediate reaction  
by-product during the first precursor gas flowing and reacting the reactive  
intermediate gas with the intermediate reaction by-product during the reactive  
intermediate gas flowing.  
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3. The method of claim 1 wherein the second precursor is different in  
composition from the first precursor.
4. The method of claim 1 wherein there is a period of time between  
25 the first precursor gas flowing and the reactive intermediate gas flowing during  
which no gas is fed to the chamber.
5. The method of claim 1 wherein there is a period of time between  
the reactive intermediate gas flowing and the second precursor gas flowing during  
30 which no gas is fed to the chamber.

6. The method of claim 1 wherein,

there is a first period of time between the first precursor gas flowing and the reactive intermediate gas flowing during which no gas is fed to the chamber; and

5 there is a second period of time between the reactive intermediate gas flowing and the second precursor gas flowing during which no gas is fed to the chamber.

7. The method of claim 1 wherein the reactive intermediate gas  
10 comprises  $\text{Cl}_2$ .

8. The method of claim 1 wherein the reactive intermediate gas comprises  $\text{H}_2$ .

9. The method of claim 1 wherein the reactive intermediate gas is not  
15 capable of reaction with the first precursor under said conditions.

10. The method of claim 1 wherein the reactive intermediate gas is not  
20 capable of reaction with the second precursor under said conditions.

11. The method of claim 1 wherein the first precursor comprises  $\text{TiCl}_4$ , the second precursor comprises  $\text{NH}_3$ , and the reactive intermediate gas comprises at least one of  $\text{H}_2$  and  $\text{Cl}_2$ .

12. The method of claim 1 wherein the first precursor comprises  $\text{NH}_3$ , the second precursor comprises  $\text{TiCl}_4$ , and the reactive intermediate gas comprises at least one of  $\text{H}_2$  and  $\text{Cl}_2$ .

13. The method of claim 1 comprising flowing multiple time spaced inert  
30 purge gas pulses to the substrate within the deposition chamber intermediate the first precursor flowing and the second precursor flowing.

14. The method of claim 1 wherein the first precursor gas flowing, the reactive intermediate gas flowing and the second precursor gas flowing collectively  
35 comprise multiple time spaced gas pulses to the substrate within the deposition chamber.

15. The method of claim 14 comprising a total period of time between the first precursor feeding and the second precursor feeding, the total period of time being characterized by no gas being fed to the chamber during said total period but for said reactive intermediate gas flowing.

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16. The method of claim 1 wherein the reactive intermediate gas flowing is plasma enhanced.

17. The method of claim 1 wherein the reactive intermediate gas flowing is plasma enhanced by plasma generation within the chamber.

18. The method of claim 1 wherein the reactive intermediate gas flowing is plasma enhanced by plasma generation remote of the chamber.

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19. An atomic layer deposition method, comprising:

positioning a semiconductor substrate within an atomic layer deposition chamber;

flowing a first precursor gas to the substrate within the atomic layer deposition chamber effective to form a first monolayer on the substrate;

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after forming the first monolayer, flowing a reactive intermediate gas to the substrate within the deposition chamber, the reactive intermediate gas being capable of reaction with an intermediate reaction by-product from the first precursor flowing under conditions of the reactive intermediate gas flowing and not being capable of reaction with the first monolayer under said conditions; and

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after flowing the reactive intermediate gas, flowing a second precursor gas different in composition from the first precursor gas to the substrate within the deposition chamber effective to form a second monolayer on the first monolayer.

20. The method of claim 19 wherein the reactive intermediate gas is not capable of reaction with, and does not react with, any exposed portion of the substrate under said conditions during the flowing of the reactive intermediate gas.

21. The method of claim 19 wherein the reactive intermediate gas is not capable of reaction with the first precursor under said conditions.

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22. The method of claim 19 wherein the reactive intermediate gas is not capable of reaction with the second precursor under said conditions.

23. The method of claim 19 comprising forming the intermediate reaction  
5 by-product during the first precursor gas flowing and reacting the reactive intermediate gas with the intermediate reaction by-product during the reactive intermediate gas flowing.

24. The method of claim 19 comprising forming the intermediate reaction  
10 by-product during the first precursor gas flowing and reacting the reactive intermediate gas with the intermediate reaction by-product during the reactive intermediate gas flowing; and

wherein the reactive intermediate gas is not capable of reaction with, and does not react with, any exposed portion of the substrate under said conditions  
15 during the flowing of the reactive intermediate gas.

25. The method of claim 24 wherein the reactive intermediate gas is not capable of reaction with the first precursor under said conditions.

20 26. The method of claim 24 wherein the reactive intermediate gas is not capable of reaction with the second precursor under said conditions.

27. The method of claim 24 wherein the reactive intermediate gas is not capable of reaction with either of the first precursor or the second precursor under  
25 said conditions.

28. The method of claim 24 comprising flowing multiple time spaced inert purge gas pulses to the substrate within the deposition chamber intermediate the first precursor flowing and the second precursor flowing.

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29. The method of claim 24 wherein the first precursor gas flowing, the reactive intermediate gas flowing and the second precursor gas flowing collectively comprise multiple time spaced gas pulses to the substrate within the deposition chamber.

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30. An atomic layer deposition method, comprising:

positioning a semiconductor substrate within an atomic layer deposition chamber;

5 flowing a first precursor gas to the substrate within the atomic layer deposition chamber effective to form a first monolayer on the substrate;

after forming the first monolayer, flowing a reactive intermediate gas to the substrate within the deposition chamber, the reactive intermediate gas being capable of reaction with an intermediate reaction by-product from the first precursor flowing under conditions of the reactive intermediate gas flowing;

10 after flowing the reactive intermediate gas, flowing an inert purge gas to the substrate within the deposition chamber; and

after flowing the inert purge gas, flowing a second precursor gas to the substrate within the deposition chamber effective to form a second monolayer on the first monolayer.

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31. The method of claim 30 wherein the reactive intermediate gas is not capable of reaction with, and does not react with, any exposed portion of the substrate under said conditions during the flowing of the reactive intermediate gas.

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32. The method of claim 30 wherein the reactive intermediate gas is not capable of reaction with the first precursor under said conditions.

33. The method of claim 30 wherein the reactive intermediate gas is not capable of reaction with the second precursor under said conditions.

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34. The method of claim 30 comprising forming the intermediate reaction by-product during the first precursor gas flowing and reacting the reactive intermediate gas with the intermediate reaction by-product during the reactive intermediate gas flowing.

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35. The method of claim 30 wherein the inert purge gas flowing is as a pulse which is time spaced from all other gas flowings, and comprising conducting multiple inert purge gas pulses intermediate the first precursor flowing and the second precursor flowing.

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36. The method of claim 30 wherein the first precursor gas flowing, the reactive intermediate gas flowing, the inert gas flowing and the second precursor gas flowing collectively comprise multiple time spaced gas pulses to the substrate within the deposition chamber.

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37. An atomic layer deposition method, comprising:

positioning a semiconductor substrate within an atomic layer deposition chamber;

10 flowing a first precursor gas to the substrate within the atomic layer deposition chamber effective to form a first monolayer on the substrate;

after forming the first monolayer, flowing an inert purge gas to the substrate within the deposition chamber;

15 after flowing the reactive intermediate gas, flowing a reactive intermediate gas to the substrate within the deposition chamber, the reactive intermediate gas being capable of reaction with an intermediate reaction by-product from the first precursor flowing under conditions of the reactive intermediate gas flowing; and

after flowing the inert purge gas, flowing a second precursor gas to the substrate within the deposition chamber effective to form a second monolayer on the first monolayer.

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38. The method of claim 37 wherein the reactive intermediate gas is not capable of reaction with, and does not react with, any exposed portion of the substrate under said conditions during the flowing of the reactive intermediate gas.

25 39. The method of claim 37 wherein the reactive intermediate gas is not capable of reaction with the first precursor under said conditions.

40. The method of claim 37 wherein the reactive intermediate gas is not capable of reaction with the second precursor under said conditions.

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41. The method of claim 37 comprising forming the intermediate reaction by-product during the first precursor gas flowing and reacting the reactive intermediate gas with the intermediate reaction by-product during the reactive intermediate gas flowing.

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42. The method of claim 37 wherein the inert purge gas flowing is as a pulse which is time spaced from all other gas flowings, and comprising conducting multiple inert purge gas pulses intermediate the first precursor flowing and the second precursor flowing.

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43. The method of claim 37 wherein the first precursor gas flowing, the inert gas flowing, the reactive intermediate gas flowing, and the second precursor gas flowing collectively comprise multiple time spaced gas pulses to the substrate within the deposition chamber.

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44. An atomic layer deposition method, comprising:

positioning a semiconductor substrate within an atomic layer deposition chamber;

flowing a first precursor gas to the substrate within the atomic layer deposition chamber effective to form a first monolayer on the substrate;

after forming the first monolayer, flowing a plurality of inert purge gas pulses to the substrate within the deposition chamber, the plurality of inert purge gas pulses comprising at least some period of time between at least two adjacent inert purge gas pulses when no gas is fed to the chamber; and

after said plurality of inert purge gas pulses, flowing a second precursor gas to the substrate within the deposition chamber effective to form a second monolayer on the first monolayer.

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45. The method of claim 44 wherein the plurality is two.

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46. The method of claim 44 wherein the plurality is more than two.

47. The method of claim 44 wherein the plurality is characterized by a common inert purge gas composition in all of said plurality.

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48. The method of claim 44 comprising a total period of time between the first precursor feeding and the second precursor feeding, the total period of time being characterized by no gas being fed to the chamber during said total period but for said plurality of inert purge gas pulses.

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49. The method of claim 44 comprising a total period of time between said two adjacent inert purge gas pulses, the total period of time being characterized by no gas being fed to the chamber during said total period.

5 50. The method of claim 44 comprising a total period of time between said two adjacent inert purge gas pulses, the total period of time comprising flowing a reactive intermediate gas to the substrate within the deposition chamber, the reactive intermediate gas being capable of reaction with an intermediate reaction by-product from the first precursor flowing under conditions of the reactive  
10 intermediate gas flowing.

51. The method of claim 50 wherein the reactive intermediate gas is not capable of reaction with, and does not react with, any exposed portion of the substrate under said conditions during the flowing of the reactive intermediate gas.  
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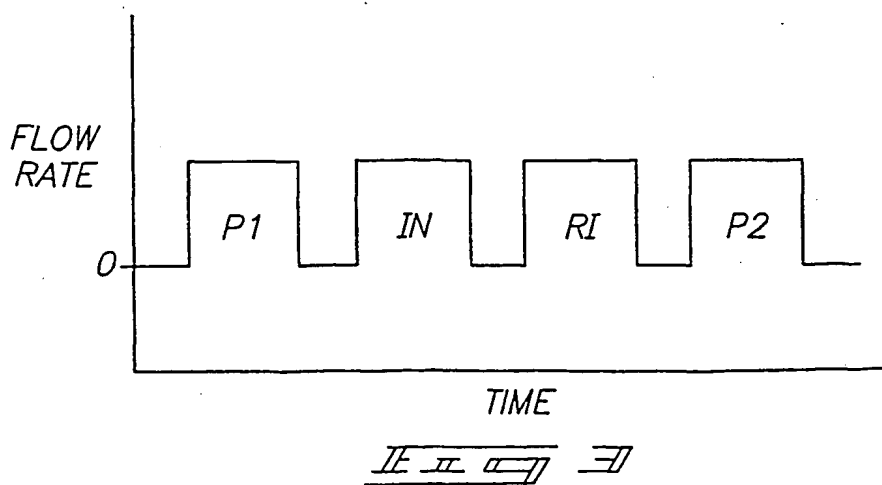
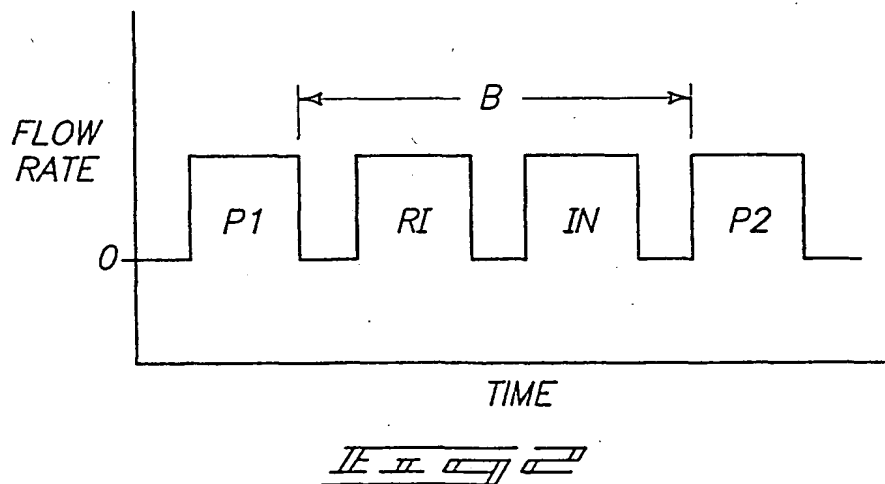
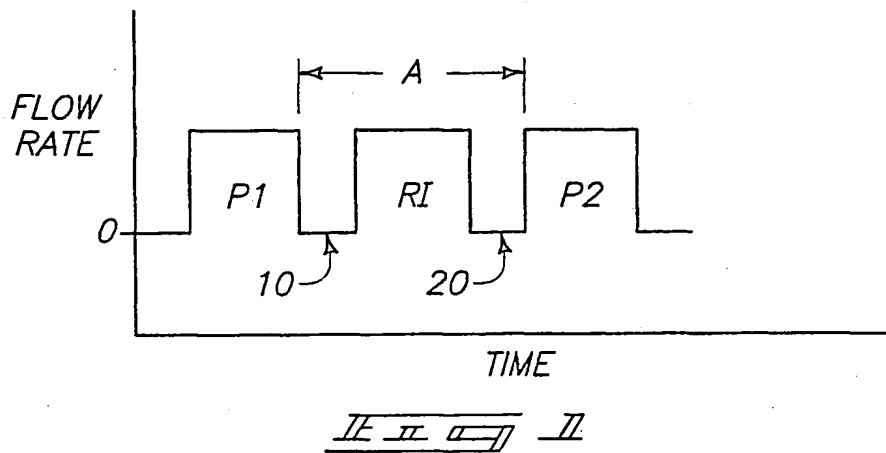
52. The method of claim 50 comprising forming the intermediate reaction by-product during the first precursor gas flowing and reacting the reactive intermediate gas with the intermediate reaction by-product during the reactive intermediate gas flowing.  
20

53. The method of claim 50 comprising forming the intermediate reaction by-product during the first precursor gas flowing and reacting the reactive intermediate gas with the intermediate reaction by-product during the reactive intermediate gas flowing; and  
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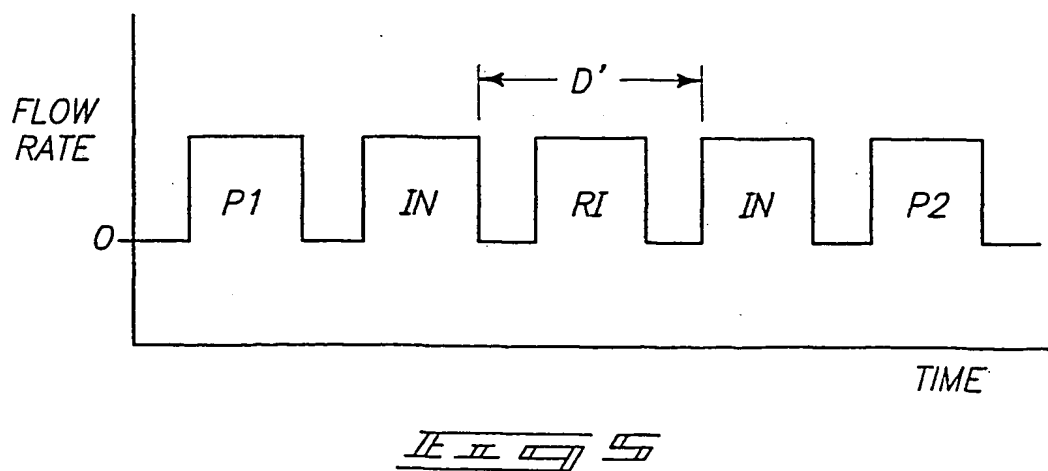
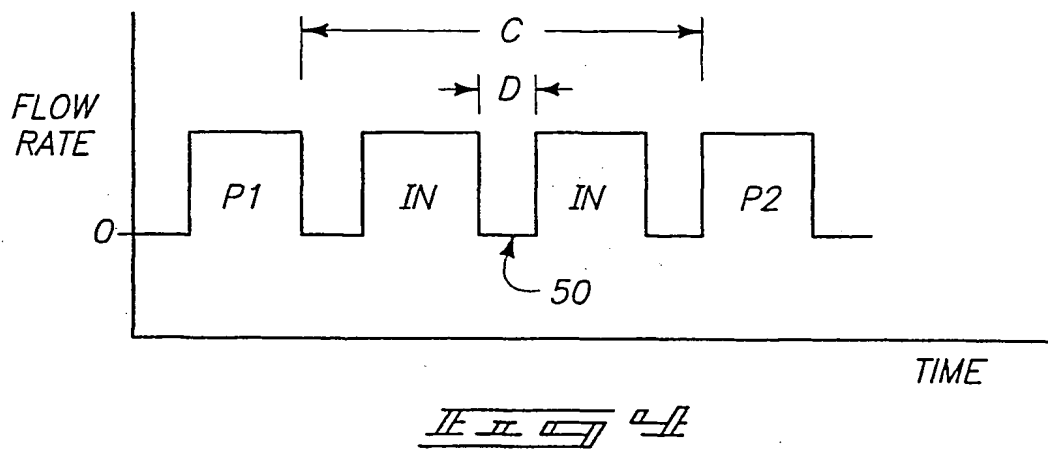
wherein the reactive intermediate gas is not capable of reaction with, and does not react with, any exposed portion of the substrate under said conditions during the flowing of the reactive intermediate gas.

54. The method of claim 44 wherein the second precursor is different in  
30 composition from the first precursor.

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## INTERNATIONAL SEARCH REPORT

International Application No

PCT/US 03/22804

A. CLASSIFICATION OF SUBJECT MATTER  
 IPC 7 C23C16/44 C30B25/14

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C30B C23C

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal

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A	WO 01 27347 A (ASM MICROCHEMISTRY OY ;HAUKKA SUVI PAEIVIKKI (FI); KAPIO SARI JOH) 19 April 2001 (2001-04-19) claims 1-6	1-54
A	US 6 200 893 B1 (SNEH OFER) 13 March 2001 (2001-03-13) claim 30	1, 19, 30, 37, 44
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Further documents are listed in the continuation of box C.



Patent family members are listed in annex.

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"&" document member of the same patent family

Date of the actual completion of the international search

1 December 2003

Date of mailing of the international search report

09/12/2003

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